A Neutron Diffraction Study of Pyridinium-1-dicyanomethylide, C₈D₅N₃, at 294 K and 118 K

By L. Devos, F. BAERT AND R. FOURET

Laboratoire de Physique des Solides, Equipe de Dynamique des Cristaux Moléculaires associée au CNRS (n° 465), Université des Sciences et Techniques de Lille, 59655 Villeneuve d'Ascq CEDEX, France

and M. Thomas

Institut Laue-Langevin, BP nº 156, 38042 Grenoble, France

(Received 24 July 1979; accepted 7 February 1980)

Abstract

neutron diffraction study of perdeuterated Α pyridinium-1-dicyanomethylide has been carried out at room temperature and at 118 K at the high-flux beam reactor of the Institut Laue-Langevin. Low temperature was obtained with a cryostat based on the Joule-Thomson expansion of gaseous helium, with a maximum long-term variation of +1 K. The crystals are monoclinic, space group $P2_1/m$. The data refinements, including extinction parameters, gave final $R_{w}(F^{2})$ values of 0.038 (RT, 552 independent reflections) and 0.028 (118 K, 1869 independent reflections). A TLS thermal analysis indicated rigid-body behaviour of the non-deuterium atoms. The non-planarity of the molecule was confirmed. The bond distances were corrected for thermal motion and their e.s.d.'s ranged from 0.0012 to 0.0030 Å (RT) and from 0.0003 to 0.0008 Å (118 K). [Crystal data: at 294 K, a =7.170 (3), b = 12.523 (5), c = 3.878 (1) Å, $\beta =$ 94.88 (1)°, $D_x = 1.418$ Mg m⁻³; at 118 K, a =7.106 (3), b = 12.461 (5), c = 3.802 (1) Å, $\beta =$ 95.47 (1)°, $D_x = 1.468 \text{ Mg m}^{-3}$.]

Introduction

The structure of pyridinium-1-dicyanomethylide (hereafter PDM) was first solved by Bugg & Sass (1965) from Weissenberg photographs. PDM belongs to the ylide family and its study is interesting from the electron-distribution point of view. This work is the neutron diffraction part of an electron density determination of PDM by the X-N method. Two series of data collections have been performed at the high-flux reactor of the Institut Laue-Langevin at Grenoble. The first, at room temperature, was for a six-day allocated time and an ω step-scan technique was chosen to save time (detector displacement would have been slower). For the second, at 118 K, the $\omega/2\theta$ step-scan technique could be used since we had 21 days for measurements.

Experimental

Deuterated crystals were grown by slow evaporation of a solution of PDM (deuterated) in N,N-dimethylformamide at room temperature. The same sample was used in the two series of measurements. It came from an as-grown small stick about 12 mm long which was cut by a dissolving saw into two halves to obtain a more suitable length of about 6 mm (beam homogeneity) and a volume of 8.2 mm³. Its colour was light vellow and it was not optically transparent, though monocrystalline. It had nine boundary planes (100), (100), (010), (010), (011), (011), (111), (111) and (305) (the last being an approximation of the sawed face) at 0.012, 2.179, 0.24, 0.54, 0.0, 0.0, 0.0, 0.0 and 5.98 mm, respectively, from the origin located at the intersection of $(01\overline{1})$, $(0\overline{1}\overline{1})$, $(\overline{1}1\overline{1})$ and $(\overline{1}1\overline{1})$. The two {010} faces were irregularly shaped due to the growth and had to be approximated by mean planes (distances indicated above) for absorption and extinction corrections.

Cell parameters were determined (X-ray diffraction, four-circle Philips PW 1100 diffractometer) by least squares from the setting angles of 25 reflections (38.0 $< \theta < 47.6^{\circ}$, Cu K α radiation, at 294 K and 16.5 $< \theta < 21.2^{\circ}$, Mo K α radiation, at 118 K). Results are listed in Table 1.* The choice of unit cell differs from that of Bugg & Sass (1965). The absorption coefficient was measured by the attenuation of a narrow beam of neutrons ($\lambda = 0.8317$ Å) passing through a deuterated

^{*} Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34932 (20 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Crystal data

Pyridinium-1-dicyanomethylide, C₂D₅N₃ $M_r = 148 \cdot 18$ Space group $P2_1/m$ Z = 2118 K 294 K a = 7.170(3) Å *a* = 7·106 (3) Å b = 12.461(5)b = 12.523(5)3.802 (1) c = 3.878(1)c = $\beta = 94.88(1)^{\circ}$ $\beta = 95.47(1)^{\circ}$

 $U = 346.942 \text{ Å}^3$ $D_x = -1.418 \text{ Mg m}^{-3}$

PDM crystal of known thickness and resulted in a value of 0.031 mm^{-1} .

 $U = 335 \cdot 126 \text{ Å}^3$

 $D_{r} = -1.468 \text{ Mg m}^{-3}$

Data were collected on the D9 four-circle diffractometer of the Institut Laue-Langevin. In the two series of measurements the crystal was mounted with c close to, but not along, the φ axis of the diffractometer. The low temperature was obtained using a Displex cooler, from Air Products and Chemical Inc., with an observed stability of ± 1 K (long-term variations) for the 21 d of measurements. In the room-temperature measurements the intensity I of the unique standard reflection showed a slow and continuous decrease of about 0.04I, due to a small variation in the alignment of the crystal. A fraction of the collected data were then remeasured with an updated orientation matrix. No such behaviour was observed for the two standards at 118 K. The F_a^2 values were obtained with the minimal $\sigma(I)/I$ criterion (Lehmann & Larsen, 1974).

Room-temperature data collection

Two octants of the Ewald sphere were measured (hboth positive and negative) as well as a small part (132) reflections) with k negative (k = -1 and -2) giving 800 reflections (sin $\theta/\lambda \leq 0.65$ Å⁻¹) comprising Friedel pairs (in the zero layer) and doubly measured ones. After averaging, and discarding four data, 552 unique reflections were left for the refinement.

118 K data collection

Measurements also covered two octants but with -21 < k < 4 and sin $\theta/\lambda \le 0.877$ Å⁻¹. 2231 reflections were measured giving 1885 data after averaging (338 equivalent reflections). 16 data were discarded leaving 1869 reflections for the refinement.

Refinement of the structure

Data were corrected for absorption by the Gaussiangrid integration method (Coppens, Leiserowitz & Rabinovich, 1965). The 26 coordinates and 52 aniso-

Table 2. Final fractional coordinates (×10	Table 2.	Final	fractional	coordinates	(×10 [±]
--	----------	-------	------------	-------------	-------------------

294 K			
	x	У	z
C(1)	95418 (25)	-25000	-13950 (56)
C(2)	86717 (16)	-34500 (10)	-6740 (40)
C(3)	69975 (15)	-34344 (8)	7747 (36)
C(4)	44361 (21)	-25000	30469 (51)
C(5)	36158 (16)	-34596 (9)	38879 (39)
N(1)	61605 (16)	-25000	15166 (36)
N(2)	28971 (14)	-42429 (7)	46504 (35)
D(1)	108819 (32)	-25000	-25149 (72)
D(2)	92760 (23)	-42090 (12)	-12469 (54)
D(3)	62822 (21)	-41553 (9)	14372 (49)
118 K			
C(1)	96617 (6)	-25000	-16018 (14)
C(2)	87753 (5)	-34601 (3)	-8690 (11)
C(3)	70697 (5)	-34473 (2)	5985 (10)
C(4)	44856 (6)	-25000	28745 (14)
C(5)	36626 (5)	-34694 (3)	37661 (10)
N(1)	62312 (4)	-25000	13448 (9)
N(2)	29362 (4)	-42590 (2)	45787 (9)
D(1)	110003 (8)	-25000	-27386 (21)
D(2)	93799 (7)	-42301 (4)	-14607 (15)
D(3)	63511 (6)	-41761 (3)	12655 (14)

Table 3. Mosaic-spread distribution

Mosaic	tensor × 10) ⁸ (Thornley	& Nelmes.	1974)			
		R ₁₁	R 22	R 33	R ₁₂	R 13	R_{23}
294	к	4.4 (2)	7.5 (6)	9.1 (3)	1.5 (2)	-1.2(4)	2.3 (8)
118	к	67 (3)	219 (10)	144 (4)	-18 (4)	-1(2)	-26 (5)
Principa	l axes and	directions					
	:	294 K				118 K	
η_i	Di	rection cosi	nes*	η_i	E	Direction co	osines*
6.4"	0.92	-0.38	-0.01	26-4	″ 0·99	0.11	-0.03
8.2	0.30	0.71	0.64	49.8	-0.11	0.94	-0.31
10.8	-0.23	-0.59	0.77	38-1	-0.01	0.31	0.95

* Direction cosines are given relative to orthogonal angström space along a, b and c.

10.8

tropic temperature factors of the atoms, the scale factor and the 6 anisotropic type 1 extinction parameters [Lorentzian distribution; Thornley & Nelmes (1974) model for the mosaic distribution] were refined by least squares, including all the data (even those with $F_o^2 \leq 0$), with *LINEX* (Becker & Coppens, 1975). The minimized quantity was $\sum w |F_o^2 - k^2 F_c^2|^2$ with $w^{-1} =$ $\sigma^2_{\text{counting}}(F_o^2).$

The resulting atomic parameters are reported in Table 2, the extinction parameters in Table 3 and the final agreement factors in Table 4. The neutron scattering lengths were (in fm) $b_{\rm C} = 6.648$, $b_{\rm N} = 9.40$ and $b_{\rm D} = 6.672$ (Shull, 1972). An additional refinement of these quantities yielded on average $b_{\rm C}$ = 6.670 (20), $b_{\rm N} = 9.384$ (20), $b_{\rm D} = 6.639$ (23) and $b_{\rm C}$ = 6.669 (8), $\ddot{b}_{N} = 9.381$ (8), $b_{D} = 6.636$ (9) fm for the 294 and 118 K data sets respectively. Difference syntheses showed residual peaks of 0.4 (294 K) and

	294 K	118 K
Wavelength [Cu (200) monochromator Laue case]	0·7414 (20) Å	0·8317 (20) Å
Scan width $\Delta\theta$ variable with θ		
Range of variation of $\Delta \theta$	1·7, 2·04°	2·24, 3·24°
$R_1 = \sum_{\text{all}} F_o - k F_c /\sum_{\text{all}} F_o$	0.044	0.030
$R_2 = \left[\sum_{\text{all}} w' F_o - k F_c ^2 / \sum_{\text{all}} w' F_o^2\right]^{1/2}$	0.019	0.015
$R_{3} = \sum_{all} F_{o}^{2} - (kF_{c})^{2} / \sum_{all} F_{o}^{2}$	0.033	0-023
$R_{4} = \left[\sum_{\text{all}} w F_{o}^{2} - (kF_{c})^{2} ^{2} / \sum_{\text{all}} wF_{o}^{4}\right]^{1/2}$	0.038	0.028
$\left(\sum_{\text{aff}} w F_o - kF_c ^2 / (N_o - N_v)\right)^{1/2}$	1.96	1.81
Transmission-factor range	0.969-0.988	0.965-0.988

Table 4. Additional experimental information andrefinement results

•		
Transmission-factor range	0.969-0.988	0.965-0.988
Extinction factors y less than	$y_{300} = 0.88$	$y_{1\bar{1}\bar{1}} = 0.75$
$0.9 \ (F_{\rm corr}^2 = F_o^2/y)$	$y_{020} = 0.79$	
	$y_{060} = 0.85$	$y_{1,\bar{1}\bar{1},1} = 0.84$
	$y_{0,10,0} = 0.00$ $y_{111} = 0.53$	$y_{202} = 0.80$
	$y_{1,11,1} = 0.84$	
	$y_{131} = 0.87$	$y_{0,\tilde{1}\tilde{2},0} = 0.88$
	$V_{22} = U_{22} = 0$	

= 0.75

Y202



Fig. 1. PDM 30% probability thermal ellipsoids at (a) 294 K and (b) 118 K (ORTEP, Johnson, 1965).

1.1 fm Å⁻³ (118 K) to be compared to Fourier synthesis peaks of 82.4 (3010 independent F_c) and 140.2 fm Å⁻³ (1869 independent F_c) respectively, corresponding to N(1) (see Fig. 1 for numbering). Comparison of the values of the mosaic-spread tensors Z (Table 3) for the two temperatures, as well as those of the y extinction factors (Table 4), shows an important modification of the mosaic distribution upon cooling of the crystal, making it less sensitive to extinction. At the same time, the widths of the rocking curves of the reflections were larger in the low-temperature measurements [e.g. 0.25° full-width for the 202 reflection at 294 K and 0.4° at 118 K].

Thermal motion

The T, L and S tensors of the rigid-body model of Schomaker & Trueblood (1968) were fitted to the U_{ii} values of the C and N atoms, Table 5. The small r.m.s. ΔU_{ii} values (0.0015 Å² at 294 K and 0.0006 Å² at 118 K) indicate that these atoms form a reasonable rigid body. However, the deuterium ΔU_{ii} values calculated from these T, L and S tensors are one order of magnitude greater than those of the heavier atoms, proving that these atoms are not rigidly linked to the C and N atom skeleton. In Table 5, eigenvalues and eigenvectors of the inertia tensor I and of the L and T tensors are also reported. They show that the eigenvectors of L and T are close to those of I (especially for L) and that the largest libration is about the axis of smallest inertia. Corrections to bond lengths were obtained from the rigid-body librations. For the D atoms, an additional $\Delta d_{\rm riding}$ correction based on the riding model (Busing & Levy, 1964) was added to $\Delta d_{rigid-body}$ so that the total correction was $\Delta d = \Delta d_{\text{rigid-body}} + \Delta d_{\text{riding}}$. Δd_{riding} was obtained from the residual tensor $\Delta U_{ij} = (U_{ij})_o - (U_{ij})_{\text{rigid-body}}$ for the D atoms and $\Delta U_{ij} = 0$ for other atoms (Johnson, 1970). The corrected distances are reported in Fig. 2. Δd_{riding} values were found to be 0.0244, 0.0193 and 0.0202 Å at 294 K and 0.0159, 0.0120 and 0.0133 Å at 118 K for D(1), D(2) and D(3) respectively.



Table 5. Rigid-body analysis

Calculations were based on non-deuterium atom U_{ii} orthogonal ångström space along a, b and c*.

294 K						
Trans	slation ten $(\dot{A}^2 \times 10^4)$	sor T)		Librai (r	tional tens ad² × 104	or L)
380 (6)	0 261 (7)	-41 (0 364 (7) 9) Cross tensor (Å rad × 10	89 (5) • s • ⁵)	0 21 (3)	-26 (3) 0 41 (2)
	2	0 28 (25) 0	-75 (44) 0 133 (24)	0 31 (20 0))	

The origin is at (x,y,z) = (0.5959, -0.25, 0.1775). The r.m.s. difference between observed and calculated U_{ii} values is 0.0015 Å². The e.s.d. of U_{ij} values is 0.0018 Å².

	Eigenvalues	Eigenvectors†		
I (tensor of inertia)	135 (at. wt Ų) 297 432	0·9049 0·0 0·4256	$\begin{array}{ccc} 0{\cdot}0 & -0{\cdot}4256 \\ 1{\cdot}0 & 0{\cdot}0 \\ 0{\cdot}0 & 0{\cdot}9049 \end{array}$	
L	$ \begin{array}{l} 100 (rad^2 \times 10^{-4}) \\ 21 \\ 30 \end{array} $	0·9192 0·0 0·3937	$\begin{array}{ccc} 0{\cdot}0 & -0{\cdot}3937 \\ 1{\cdot}0 & 0{\cdot}0 \\ 0{\cdot}0 & 0{\cdot}9192 \end{array}$	
Т	413 ($\dot{A}^2 \times 10^{-4}$) 261 330	0·7722 0·0 0·6354	$\begin{array}{ccc} 0.0 & -0.6354 \\ 1.0 & 0.0 \\ 0.0 & 0.7722 \end{array}$	

118 K

Transl (ational ten $\dot{A}^2 \times 10^4$)	sor T		Libr: (ational tenso rad ² × 10 ⁴)	or L
127 (3)	0 88 (3)	-9 (0 120 (3) 4)	30 (2)	0 7 (1)	-9 (1) 0 14 (1)
		(Cross tenso (Å rad × 1	or S 0 ⁵)		
	14	0 4 (10) 0	-19 (18) 0 32 (10)	0 19 (8) 0		

The origin is at (x,y,z) = (0.6029, -0.25, 0.1621). The r.m.s. difference between observed and calculated U_{ij} values is 0.0006 Å². The e.s.d. of U_{ij} values is 0.0007 Å².

	Eigenvalues	Eig	envect	ors†
I	137 (at. wt Å ²)	0·9062	0.0	-0.4228
	300	0·0	1.0	0.0
	437	0·4228	0.0	0.9062
L	34 (rad ² × 10 ⁻⁴)	0·9142	0.0	-0.4053
	7	0·0	1.0	0.0
	10	0·4053	0.0	0.9142
Т	$ \begin{array}{c} 133 \ (\text{\AA}^2 \times 10^{-4}) \\ 88 \\ 113 \end{array} $	0.8131 0.0 0.5822	0·0 1·0 0·0	-0.5822 0.0 0.8131

† Orthogonal ångström space along a, b and c*.

‡ Crystallographic fractional coordinates.



Description and discussion of the structure

The atomic notation and thermal ellipsoids are shown in Fig. 1. The dimensions of the molecule are shown in Figs. 2 and 3. Bond lengths and angles compare well with those obtained by Bugg & Sass (1965) except for the C(5)-N(2) length [1.1582 Å (uncorrected) instead of 1.13 Å in their results]. The 1.40 Å found for the C(4)-C(5) distance, intermediate between a $C(sp^3)$ - $C(sp^3)$ and a $C(sp^2)-C(sp^2)$ bond length, is confirmed. Another small difference is in the angle C(4)-C(5)-N(2) (177.95 instead of 180°). Noticeable also is the dissymmetry between the angles C(2)-C(3)-D(3) (122.40°) and D(3)-C(3)-N(1) (116.84^{\circ}) as well as C(1)-C(2)-D(2) (121.56°) and C(3)-C(2)-D(2)(118·31°).

Table 6. Least-squares planes and deviations (Å) of the atoms from the planes

Atoms used in fitting the least-squares planes are denoted by asterisks.

Plane (1) :	0-4874 (3) <i>X</i>	+ $0.0Y + 0.8732$ (2)Z = 2.3325 (21)	(294 K)
	0-4910 (1) <i>X</i>	+ $0.0Y + 0.8712$ (1)Z = 2.2421 (5)	(118 K)
Plane (II):	0.5462(12)X	+ $0.0Y$ + 0.8736 (8)Z = 2.4918 (21)	(294 K)
	0.5668(3)X	+ $0.0Y$ + 0.8238 (2)Z = 2.4462 (6)	(118 K)

X,Y,Z are atomic coordinates in Å referred to the axes a*, b and c.

	Plane (I)			Plane (11)	
	294 K	118 K		294 K	118 K
C(1)*	0.0096 (20)	0-0129 (5)	C(1)	-0.2911 (20)	-0.3868 (5)
C(2)*	-0.0029 (15)	0.0000 (4)	C(2)	-0.2302 (15)	-0.3103 (4)
C(3)*	-0.0062 (13)	-0.0056 (4)	C(3)	-0.1330 (13)	-0.1813 (4)
C(4)*	0.0077 (19)	0.0032 (5)	C(4)*	-0.0024 (19)	-0.0023 (5)
C(5)	0.0505 (14)	0.0613(4)	C(5)*	0.0026 (14)	0.0025 (4)
C(2')*	-0.0029(15)	0.0000(4)	C(5')*	0.0026 (14)	0.0025 (4)
C(3')*	-0.0062 (13)	-0.0056 (4)	N(1)	-0.0901 (13)	-0.1257 (4)
N(1)*	-0.0018(13)	-0.0002(4)	N(2)*	-0.0014(13)	-0.0013(3)
N(2)	0.0967 (13)	0.1210(3)	N(2')*	-0.0014(13)	-0.0013(3)
D(1)*	0.0257 (27)	0.0223 (8)	D(1)	-0.3818(27)	-0.4927 (8)
D(2)*	-0.0186(20)	-0.0216(6)	D(2)	-0.2490 (20)	-0.3337(6)
D(3)*	0.0072 (18)	0.0081(5)	D(3)	-0.1056 (18)	-0.1422 (5)
D(2')*	-0.0186(20)	-0.0216(6)	R.m.s.	0.0021	0.0021
D(3')*	0.0072 (18)	0.0081 (5)	devia	tion	
R.m.s.	0.0119	0.0122	of fitt	ed atoms	
devia	tion				
of fitt	ed atoms				

Table 7. The shortest intermolecular distances (Å)

Some distances, equivalent by lattice translations, are not reported [*e.g.* $C(3)_{x,y,z} \cdots C(2)_{x,y,z+1}$ is equivalent to $C(2)_{x,y,z} \cdots C(3)_{x,y,z-1}$].

	294 K	118 K
$C(2)\cdots C^{i}(3)$	3.429 (2)	3.3531 (5)
$C(3)\cdots C^{l}(5)$	3.451 (2)	3.3781 (5)
$C(2')\cdots C^{1}(3')$	3.429 (2)	3.3531 (5)
$C(3')\cdots C^{1}(5')$	3.451(2)	3.3781 (5)
$C(5)\cdots D^{ii}(1)$	2.774(3)	2.6988 (7)
$C(5')\cdots D^{ii}(1)$	2.774(3)	2.6988 (7)
$N(2) \cdots D^{iii}(3')$	2.553(2)	2.5304 (5)
$N(2')\cdots D^{iv}(2')$	2.553(2)	2.5304 (5)
$N(2)\cdots D^{v}(2')$	2.753 (2)	2.6994 (6)
Symmetry code		
(i) $x, y, z - 1$	(iv)	$1 - x, y + \frac{1}{2}, 1 - z$
(ii) $x - 1, y, z + 1$	(v)	$1 - x, y - \frac{1}{2}, -z$
(iii) $1-x, y-\frac{1}{2}, 1-z$		· •

The non-planarity of the molecule, presumably related to a residual electronic charge on C(4), is confirmed by an angle of $2 \cdot 38^{\circ}$ between C(4)—C(5) and the best least-squares plane through C(4) and the pyridinium ring (294 K). Table 6 gives the least-squares analysis of the planarity of these last atoms as well as the best plane through C(4), C(5), N(2), C(5') and N(2'). The two planes make an angle of $5 \cdot 12^{\circ}$ (118 K) whereas a value of $3 \cdot 94^{\circ}$ is found for the roomtemperature data. Short intermolecular atomic distances are reported in Table 7.

X-ray data at the same temperatures have already been collected and their study is at present in progress.

The authors acknowledge helpful discussions with Dr M. S. Lehmann. Mr M. Muller and Mrs C. Carpentier have been of great assistance in growing the crystals.

References

- BECKER, P. J. & COPPENS, P. (1975). Acta Cryst. A31, 417-425.
- BUGG, C. & SASS, R. L. (1965). Acta Cryst. 18, 591-594.
- BUSING, W. R. & LEVY, H. A. (1964). Acta Cryst. 17, 142–146.
- COPPENS, P., LEISEROWITZ, L. & RABINOVICH, D. (1965). Acta Cryst. 18, 1035–1038.
- JOHNSON, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
- JOHNSON, C. K. (1970). Proceedings of the 1969 International Summer School on Crystallograhic Computing, edited by F. R. AHMED, pp. 224–225. Copenhagen: Munksgaard.
- LEHMANN, M. S. & LARSEN, F. K. (1974). Acta Cryst. A30, 580–584.
- SCHOMAKER, V. & TRUEBLOOD, K. N. (1968). Acta Cryst. B24, 63-76.
- SHULL, C. G. (1972). Private communication.
- THORNLEY, F. R. & NELMES, R. J. (1974). Acta Cryst. A 30, 748–757.

Acta Cryst. (1980). B36, 1811–1819

Structure of Digoxin

BY KUANTEE GO AND GOPINATH KARTHA

Roswell Park Memorial Institute, Buffalo, New York 14263, USA

AND JAMES P. CHEN

University of Tennessee Memorial Research Center, Knoxville, Tennessee 37920, USA

(Received 4 September 1979; accepted 17 March 1980)

Abstract

The three-dimensional structure of digoxin, a cardiac glycoside, was determined by X-ray diffraction analysis. Digoxin ($C_{41}H_{64}O_{14}$) crystallizes in the triclinic system, space group P1, with cell parameters a = 7.404 (2), b = 12.781 (2), c = 12.677 (2) Å, a = 91.15 (1), $\beta = 119.89$ (2) and $\gamma = 104.78$ (2)° ($\lambda = 1.5418$ Å), V = 990 Å³, Z = 1. The structure was

solved by vector-search methods using digoxigenin as known fragment followed by tangent-refinement recycling procedures. Refinement was carried out by blockdiagonal least-squares calculations to an R factor of $4 \cdot 2\%$. The molecule has an extended structure with its aglycone moiety in the 3β , 12β , 14β -trihydroxy- 5β -card-20(22)-enolide conformation and the three digitoxoses joined to the aglycone in a series of $\beta(1 \rightarrow 4)$ -Dglycosidic linkages. The first and second digitoxoses

0567-7408/80/081811-09\$01.00 © 1980 International U

© 1980 International Union of Crystallography